

Naming Chemical Compounds Review:

For each of the following identify whether it's ionic or covalent, then name it appropriately.

1) NO_2		
2) NaBr		
3) AsBr_3		
4) N_2S		
5) CaO		
6) SO_3		
7) $\text{Be}(\text{NO}_3)_2$		
8) PH_3		
9) SiO_2		
10) BF_3		
11) FeCl_2		
12) SF_6		
13) Li_2S		
14) P_2Br_4		
15) CuCl_2		
16) MgBr_2		
17) $\text{Be}(\text{OH})_2$		
18) CO		

Name:	Ionic or Covalent:	Formula:
1. Sodium phosphide		
2. Antimony tribromide		
3. Hexaboron silicide		
4. Magnesium nitrate		
5. Lead (II) sulfite		
6. Chlorine dioxide		
7. Silver oxide		
8. Calcium phosphate		
9. Gallium nitride		
10. Dinitrogen trioxide		
11. Beryllium chloride		
12. Iodine pentafluoride		
13. Barium bromide		
14. Aluminum chloride		
15. Carbon dioxide		
16. Lead (IV) nitrate		
17. Tetraphosphorus hexoxide		
18. Sodium oxide		
19. Dichlorine heptaoxide		
20. Calcium oxide		
21. Iron (III) fluoride		
22. Dihydrogen monoxide		