Covalent and Ionic Bonding

For the following,

- a) Indicate if it is an ionic or covalent bond
- b) Draw the transfer or sharing of electrons
- c) Write the resulting formula
- 1. Chlorine and Sulfur

2. Sodium and Fluorine

3. Chlorine and chlorine

4. Calcium and oxygen

2. Magnesium and Bromine			
	·		
6, Sulfur and Flüorine			
		•	
7. Magnesium and Chlorine			
3. Potassium and Oxygen			
			-
Oxygen and Chlorine			
3. Carbon and Fluorine	·		- 6.55
			•

8.

9.

10.

11: Beryllium and Nitrogen